

Pure Water for Laboratory

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Why Water Quality Matters

Many impurities can be present in water, which may interfere with testing and lead to inaccurate results

 Examples: Effects on chemical reactions, cell culture, analytical techniques

Water Purity Classes

- Laboratory water is classified based on purity levels: Type I, II, III
- Each type serves different purposes depending on purity requirements

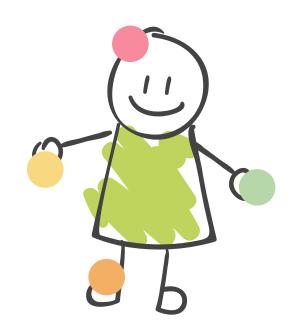
Water Purity Classes

Type I Water: Ultrapure Water

Type II Water: Pure Water

Type III Water: General Laboratory

Water



Type I Water: Ultrapure Water

Highest Purity:

- Having the highest level of purity
- Extremely low levels of ions, organics, particulates, bacteria, and pyrogens.

Resistivity

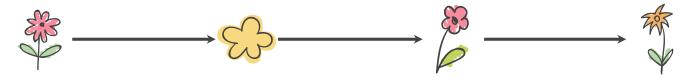
 ≥ 18.2 megohmscm (MΩ·cm) at 25°C, indicating very low ionic content Total Organic Carbon (TOC)

 Less than 5 parts per billion (ppb), signifying minimal organic contamination

Bacteria and
Particulate Content

 Virtually free of bacteria and particulates, making it suitable for highly sensitive experiments

Production Methods of Type I Water



Step 1

Reverse Osmosis (RO): Initial purification step to remove most dissolved solids, organic molecules, Step 2

Deionization (DI): Use ion exchange resins to remove cations and anions Low conductivity Step 3

Ultrafiltration : Removes particulates and pyrogens Step 4

UV Oxidation:
Ultraviolet (UV) treatment
at 185 nm and 254 nm to
degrade and oxidize
organic contaminants



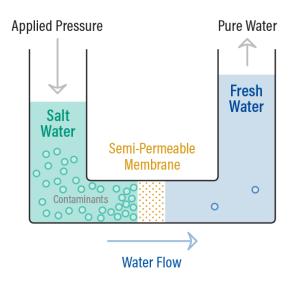
Final Filtration:

A 0.2-micron filter or finer is used to remove any remaining particulates and bacteria

Step 1: Reverse Osmosis

เป็นกระบวนการที่เกิดขึ้นเมื่อเราใส่ความดันให้กับฝั่งที่มี ความเข้มข้นของสารละลายสูง จนความดันน้ำสูงกว่า ความดันออสโมติก(Osmotic Pressure)โดยโมเลกุล ของเหลวหรือน้ำจะถูกบีบให้ผ่าน เยื่อเมมเบรน (Membrane)จากบริเวณที่มีความเข้มข้นของ สารละลายสูง ไปยังบริเวณที่มีความเข้มข้นของ สารละลายต่ำ

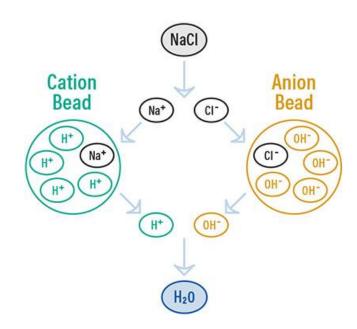
Reverse Osmosis



Step 2 : DI (Deionized water)

น้ำ DI (Deionized Water) คือ การทำน้ำให้มีความ บริสุทธิ์สูงมาก โดยผ่าน Ion Exchange เพื่อจับทั้ง ไอออนบวกและลบออกจากน้ำ

โดยกระบวนการของน้ำ DI มักจะต้องใช้ระบบ RO เพื่อ ทำให้น้ำบริสุทธิ์ในระดับนึงก่อน แล้วจึงใช้ Ion Exchange จับประจุทั้งบวกและลบอีกครั้ง เพื่อให้น้ำ ปราศจากแร่ธาตุและสิ่งเจือปนให้ได้มากที่สุดก่อน นำไปใช้งาน



Applications

- Type I water is essential for applications requiring the utmost purity
 - High-Performance Liquid Chromatography (HPLC)
 - Gas Chromatography-Mass Spectrometry (GC-MS)
 - Inductively Coupled Plasma Mass Spectrometry (ICP-MS)
 - * Molecular Biology (e.g., PCR, DNA sequencing) Cell Culture

Type II Water: Pure Water

Intermediate Purity

 Suitable for many general laboratory applications

Resistivity

Resistivity of 1-10 MΩ·cm at 25°C , Sufficient for routine analytical procedures

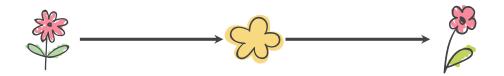
Total Organic Carbon (TOC)

Below 50 ppb

Bacteria and Particulate Content

Contains more bacteria and particulates than Type I water

Production Methods of Type II Water



Step 1

and organic molecules

Reverse Osmosis (RO): Removes dissolved salts Step 2

Deionization (DI): Removes most cations and anions Step 3

UV Treatment:
Reduces organic
contaminants and
controls microbial
content

Applications

- Type II water is used in applications that require pure water but do not demand the extreme purity of Type I water, such as:
 - Buffer Preparation
 - PH Solution Preparation
 - General Chemistry Reactions
 - Microbiological Media Preparation
 - Clinical Analyzer Feed Water

Type III Water: General Laboratory Water

Basic Purity

 the lowest purity among the three types and is intended for noncritical laboratory tasks

Resistivity

Typically has a resistivity of 0.05-1 MΩ·cm at 25°C, reflecting higher levels of ionic and particulate contaminants compared to Types I and II.

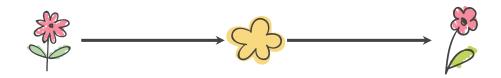
Total Organic Carbon (TOC)

Higher TOC levels are acceptable, making this water type suitable for less sensitive applications

Bacteria and
Particulate Content

Higher bacterial and particulate content is permissible

Production Methods of Type III Water



Step 1

Filtration:
Removes large
particulates and some
microbial content

Step 2

Reverse Osmosis (RO): Single-pass treatment reduces contaminants

Step 3

Carbon Filtration:
Removes chlorine
and organic
compounds

Applications

- Type III water is generally used for laboratory tasks that do not require high purity, such as:
 - Glassware Washing
 - Autoclave Feed Water
 - Heating Baths
 - Rinsing and General Cleaning
 - Non-critical Reagent Preparation

Quality Standards and Regulations



Monitoring and Testing Water Quality

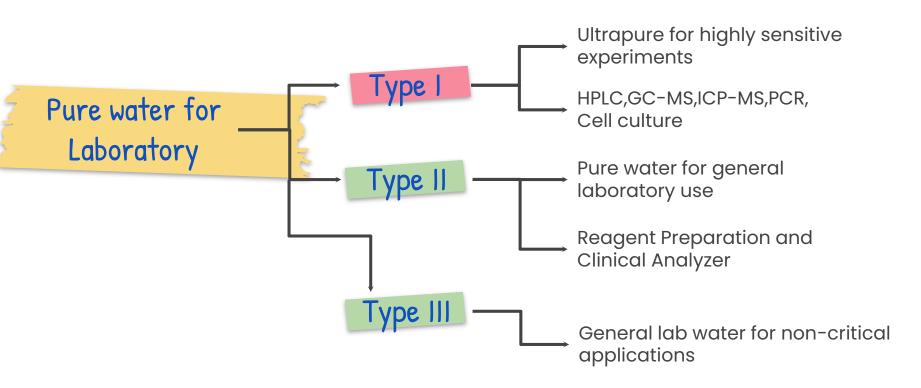
Methods

- Conductivity
- resistivity
- TOC Testing
- Microbial testing

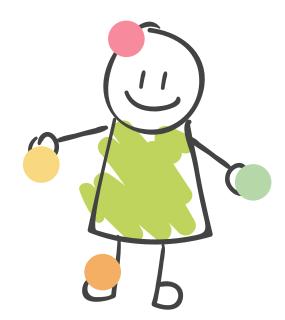
Importance

 Regular testing to maintain standards

Summary of Water Types and Their Uses



THANK YOU



75%

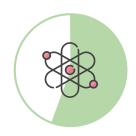
Mercury
It's the
closest to
the Sun

60%



Saturn
It's a gas giant
with many
rings

55%

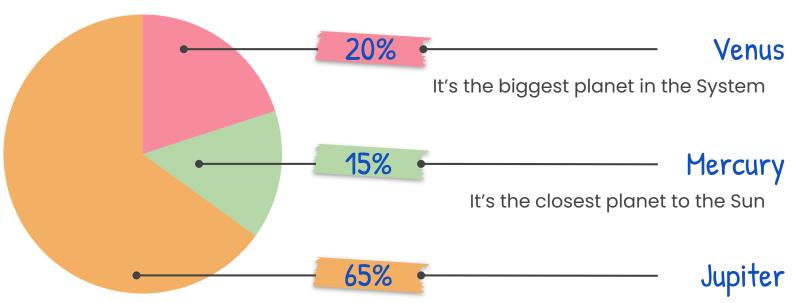


Mars
It's actually a very cold place

40%



Jupiter
It's a gas
giant and
it's very big



Follow the link in the graph to modify its data and then paste the new one here. **For more info, click here**

It's the biggest planet in the System

Monday	Tuesday	Wednesday	Thursday	Friday	Saturday	Sunday
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31	1	2	3	4



Notes

Neptune is the farthest planet from the Sun. It's also the fourth-largest object by diameter in the Solar System

Earth

It's the third planet from the Sun

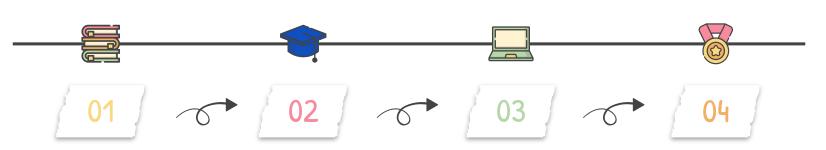
Neptune

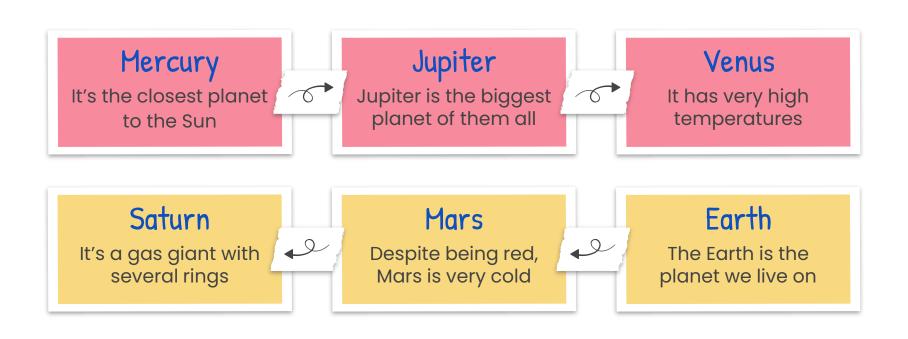
It's the farthest planet from the Sun

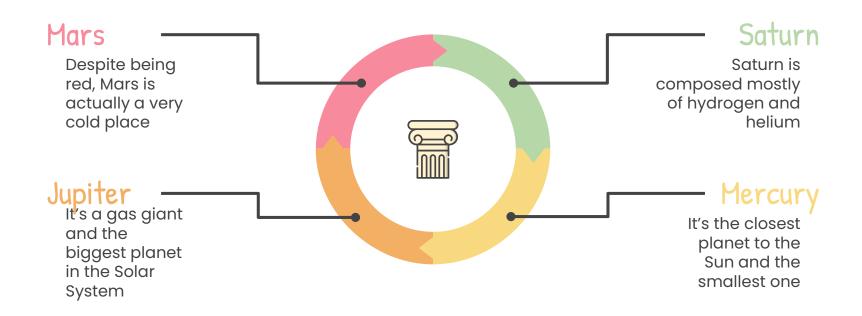
Mars

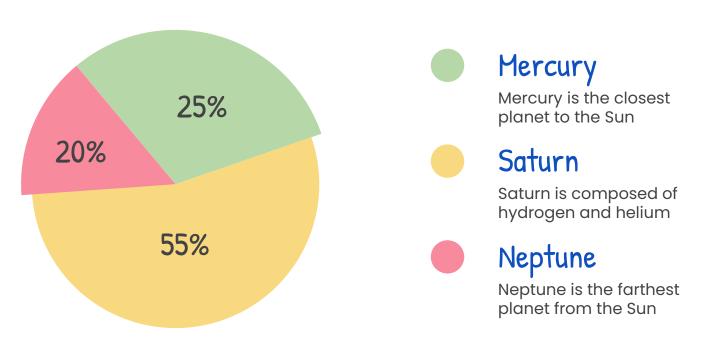
Despite being red, Mars is a cold place Mercury

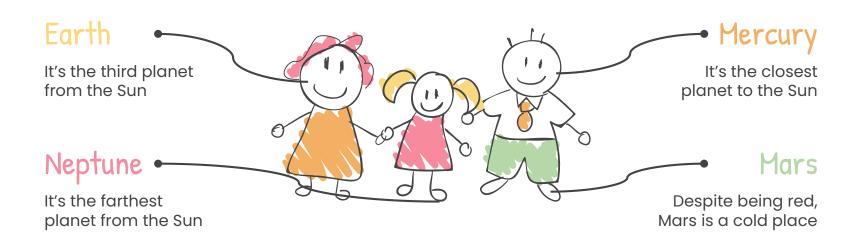
It's the closest planet to the Sun

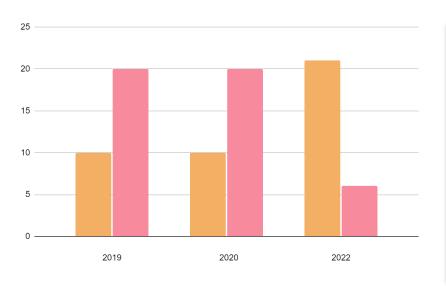






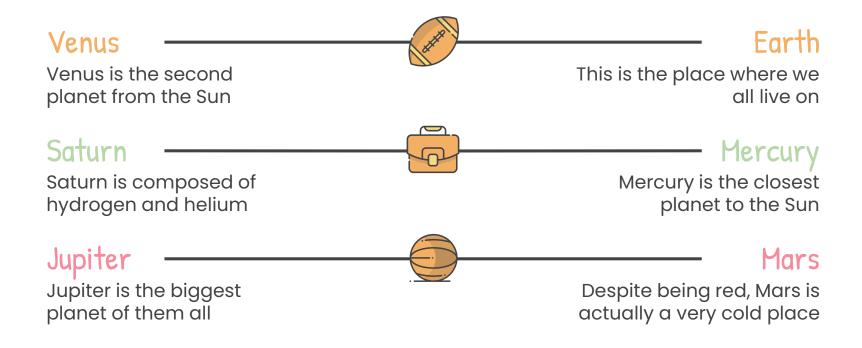


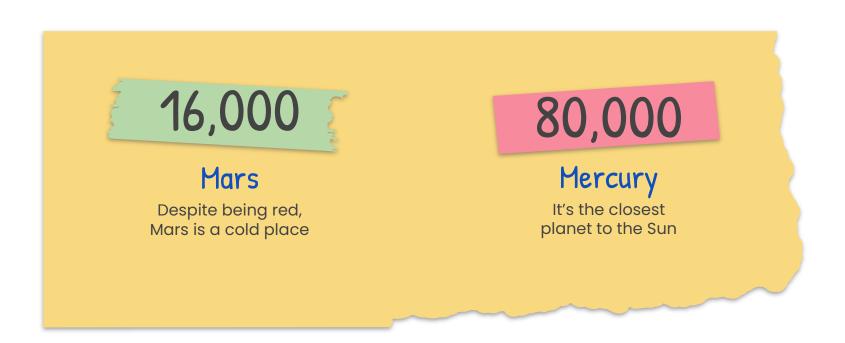


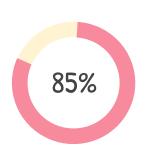




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Saturn

Saturn has several rings



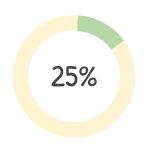
Mercury

Mercury is a very small planet



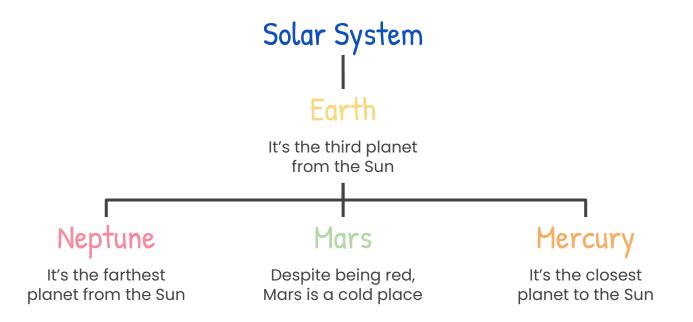
Earth

The only planet with life

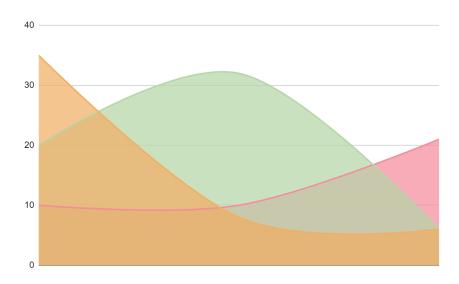


Mars

Mars is a very cold place



Mars	Venus	Earth	Neptune	
	C C C C C C C C C C C C C C C C C C C			
Mars is actually a very cold place full of iron oxide dust	Venus is the second planet from the Sun	Earth is the third planet and the one where we live on	Neptune is the farthest planet from the Sun	



Follow the link in the graph to modify its data and then paste the new one here. **For more info, click here**

Jupiter

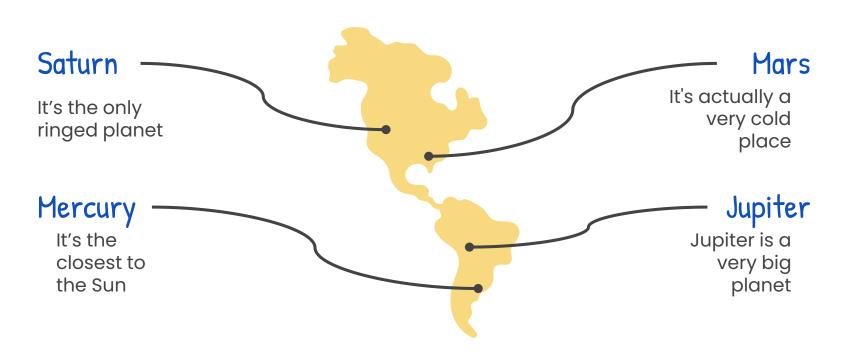
It's the biggest planet in the Solar System

Mercury

It's the smallest planet in the Solar System

Saturn

It's composed mostly of hydrogen and helium



Mercury

The closest to the Sun

Mars

A very cold place

Saturn

A ringed planet

Earth

The only planet with life









Mercury

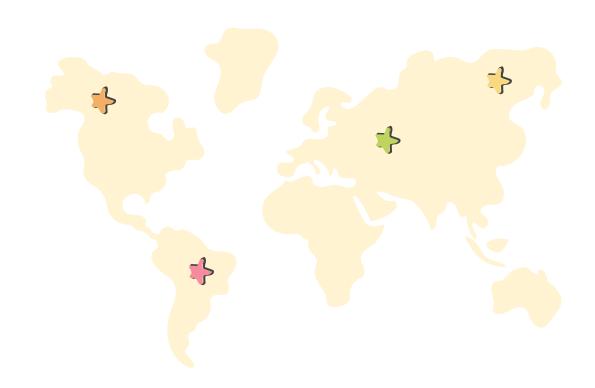
It's the closest to the Sun

Mars

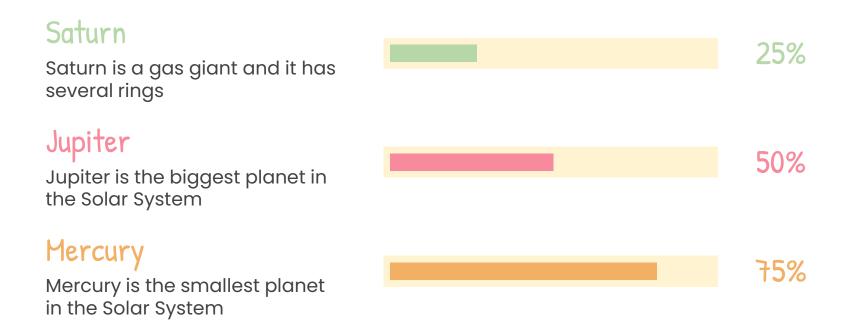
Mars is actually a cold place

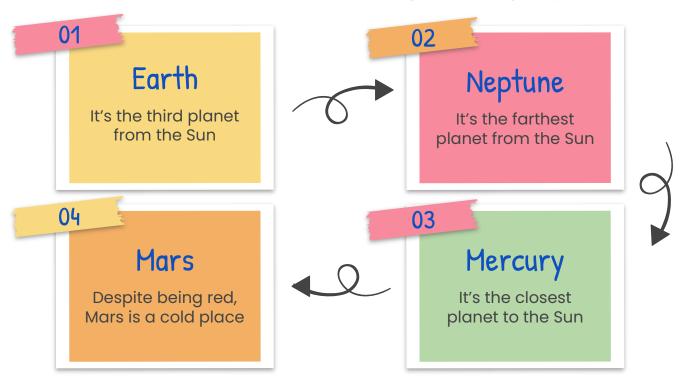
Jupiter

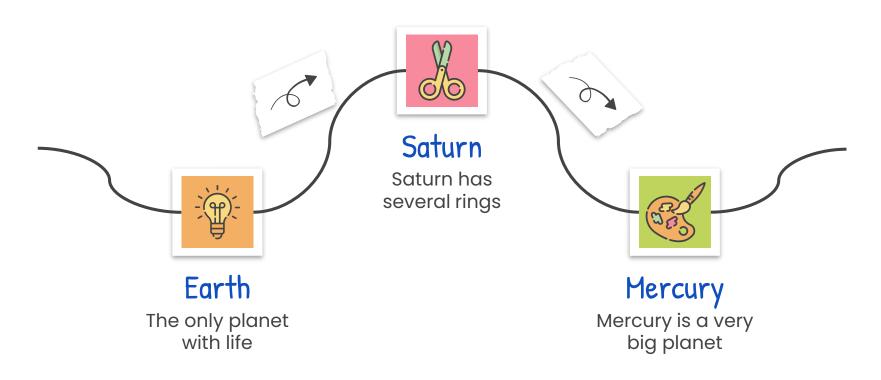
Jupiter is a very big planet



	Monday	Tuesday	Wednesday	Thursday	Friday
Week 1	◇				✓
Week 2		◆	✓		
Week 3	✓			✓	
Week 4			✓		







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